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#### Key indicators

Single-crystal X-ray study

$T = 150$  K

Mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å

$R$  factor = 0.054

$wR$  factor = 0.146

Data-to-parameter ratio = 13.0

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## (2*S*\*)-2-Hydroxy-2-[(2*S*\*,5*R*\*)-5-(1-hydroxy-1-methyl-ethyl)tetrahydrofuran-2-yl]-1-phenylethanone

The molecules of the title compound,  $\text{C}_{15}\text{H}_{20}\text{O}_4$ , link *via* a single intermolecular hydrogen bond to form chains running along the *b* axis.

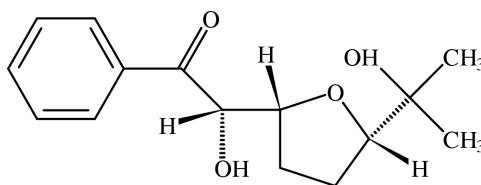
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### Comment

During the development of chiral phase-transfer catalysed permanganate oxidations of 1,5-dienes, the title compound, (I), was prepared (Brown & Keily, 2001).



(I)

An intermolecular hydrogen bond links the molecules into chains extending along the *b* axis. Also present are three intramolecular hydrogen bonds (Table 1).

### Experimental

(*E*)-7-Methyl-1-phenylocta-2,6-dien-1-one was oxidized by potassium permanganate under solid-liquid phase-transfer conditions using a chiral quaternary ammonium salt. Following a standard work-up and column chromatography, the product was recrystallized from ethanol/hexane to give colourless crystals of the title compound.

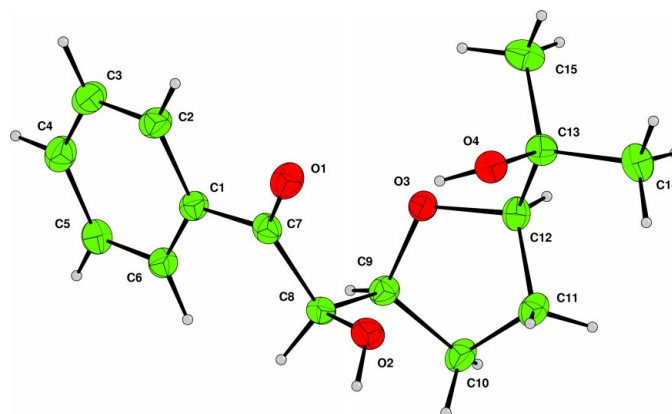


Figure 1

A view of the title compound with the atomic numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

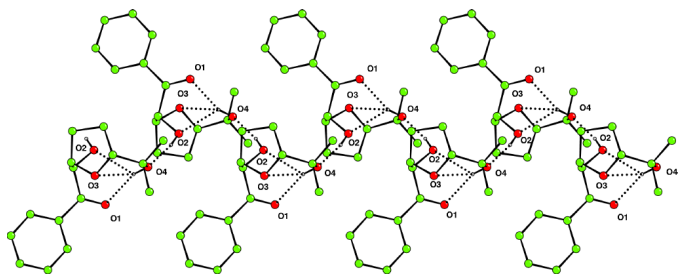


Figure 2

A view looking down the *a* axis showing the hydrogen-bonded chains extending along the *b* direction.

#### Crystal data

$C_{15}H_{20}O_4$   
 $M_r = 264.31$   
 Orthorhombic,  $P2_12_12_1$   
 $a = 7.5556$  (3) Å  
 $b = 8.0446$  (3) Å  
 $c = 22.9549$  (8) Å  
 $V = 1395.24$  (9) Å<sup>3</sup>  
 $Z = 4$   
 $D_x = 1.258$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation  
 Cell parameters from 4487 reflections  
 $\theta = 3.2$ – $25.0^\circ$   
 $\mu = 0.09$  mm<sup>-1</sup>  
 $T = 150$  (2) K  
 Block, colourless  
 $0.25 \times 0.10 \times 0.08$  mm

#### Data collection

Nonius KappaCCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans to fill the Ewald sphere  
 Absorption correction: multi-scan (SORTAV; Blessing, 1997)  
 $T_{\min} = 0.978$ ,  $T_{\max} = 0.993$   
 4487 measured reflections

2254 independent reflections  
 1917 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.062$   
 $\theta_{\max} = 25.0^\circ$   
 $h = -7 \rightarrow 8$   
 $k = -9 \rightarrow 7$   
 $l = -26 \rightarrow 22$

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.054$   
 $wR(F^2) = 0.146$   
 $S = 1.03$   
 2254 reflections  
 174 parameters

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0959P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.22$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.24$  e Å<sup>-3</sup>

Table 1

Hydrogen-bonding geometry (Å, °).

<i>D</i> –H... <i>A</i>	<i>D</i> –H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> –H... <i>A</i>
O2–H2A...O4 <sup>i</sup>	0.84	1.88	2.707 (3)	170
O4–H4A...O3	0.84	2.31	2.756 (3)	113
O4–H4A...O1	0.84	2.33	3.131 (3)	161
O4–H4A...O2	0.84	2.58	3.186 (3)	130

Symmetry code: (i)  $1 - x, y - \frac{1}{2}, \frac{3}{2} - z$ .

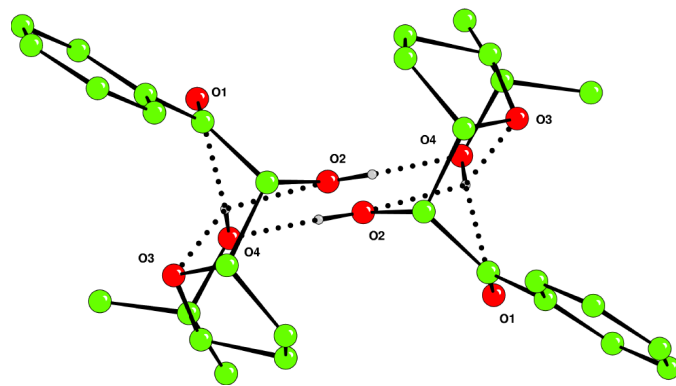


Figure 3

A view looking down the *b* axis showing both the inter- and intramolecular hydrogen bonds.

Data collection: *COLLECT* (Hooft, 1998); cell refinement: *SCALEPACK* (Otwinowski & Minor, 1997) and *COLLECT*; data reduction: *DENZO* (Otwinowski & Minor, 1997), *COLLECT* and *MAXUS* (Mackay *et al.*, 1998); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *CAMERON* (Watkin *et al.*, 1993); software used to prepare material for publication: *WINGX* (Farrugia, 1998).

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#### References

- Blessing, R. H. (1997). *J. Appl. Cryst.* **30**, 421–426.  
 Brown, R. C. D. & Keily, J. F. (2001). *Angew. Chem. Int. Ed.* In the press.  
 Farrugia, L. J. (1998). *WINGX*. University of Glasgow, Scotland.  
 Hooft, R. (1998). *COLLECT*. Nonius BV, Delft, The Netherlands.  
 Mackay, S., Gilmore, C. J., Edwards, C., Tremayne, M., Stuart, N. & Shankland, K. (1998). *MAXUS*. University of Glasgow, Scotland, Nonius BV, Delft, The Netherlands, and MacScience Co. Ltd, Yokohama, Japan.  
 Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter and R. M. Sweet, pp. 307–326. London: Academic Press.  
 Sheldrick, G. M. (1990). *Acta Cryst.* **A46**, 467–473.  
 Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.  
 Watkin, D. M., Pearce, L. & Prout, C. K. (1993). *CAMERON*. Chemical Crystallography Laboratory, University of Oxford, England.